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 $H_{2N} \xrightarrow{N}_{H} \xrightarrow{R}_{H_{2}} \xrightarrow{H_{2}} \xrightarrow{H_{2}}$

The rate of the biologically important reaction between 7,8-dihydropterins and H_2O_2 , as well as the products formed strongly depend on the chemical structure of the substituent at position 6 of the pterin moiety.

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Sergei V. Voitekhovich^{*}, Pavel N. Gaponik, Alexander S. Lyakhov, Oleg A. Ivashkevich



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Ying-Qiao Mei, Jin-Tao Liu^{*}

filig-Qiao Wei, Jili-Iao Liu



 $R_F = Per(poly)fluoroalkyl$ $R^1 = P(O)Ph_2, P(O)(OEt)_2, COOEt$

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Sabir H. Mashraqui*, Yogesh S. Sangvikar, Shailesh G. Ghadigaonkar, Mohamed Ashraf, M. Meetsma



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Valery N. Kozhevnikov*, Olga V. Shabunina, Dmitry S. Kopchuk, Maria M. Ustinova, Burkhard König, Dmitry N. Kozhevnikov*





Ph____NO2 NO₂ Ar¹⁄ ℃O₂Me 42-85% i. Toluene or MeCN, 25 °C, AgOAc or Ag_2O , 42-98% (10 examples) (22 examples) NEt₃ or DBU R= alkyl/aryl/heteroaryl

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*Corresponding author ()+ Supplementary data available via ScienceDirect

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